

Solutions Assignment 7

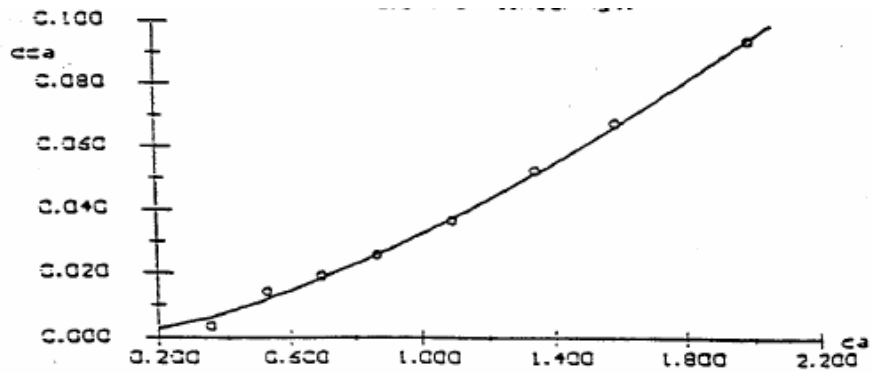
Problem 1.

$$\frac{dC_A}{dt} = kC_A^n$$

From a non-linear fit, we get

$$k = 0.033$$

$$n = 1.54$$



Model: $C_A^n = k \cdot (C_A)^n$

$$k = 0.0326929$$

$$n = 1.53995$$

4 positive residuals, 4 negative residuals. Sum of squares = $1.49367e-05$

Problem 2

t(min)	10	20	30	40	50	60
C _A (ppm)	2.45	1.74	1.23	0.88	0.62	0.44

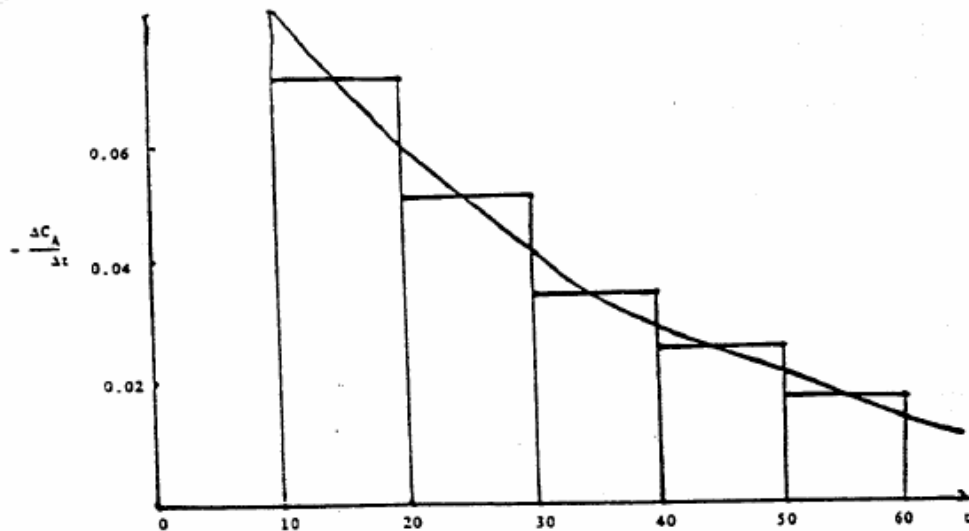
(a) Mole Balance: constant V

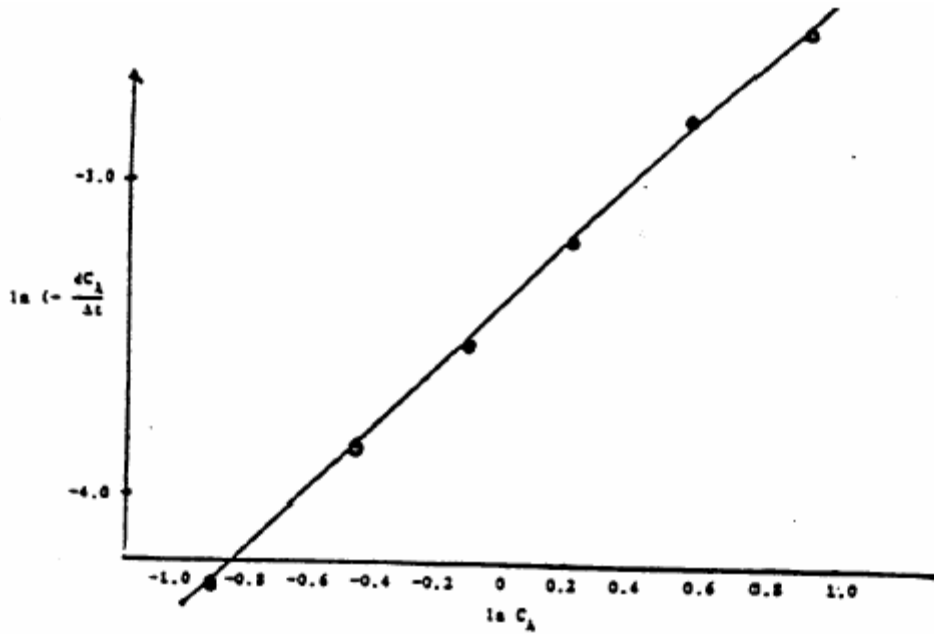
$$\frac{dC_A}{dt} = r_A = -k C_A^a$$

$$\ln \left(-\frac{dC_A}{dt} \right) = \ln k + a \ln C_A$$

Differentiation

t(min)	10	20	30	40	50	60
Δt(min)		10	10	10	10	10
C _A (ppm)	2.45	1.74	1.23	0.88	0.62	0.44
ΔC _A (ppm)		-0.71	-0.51	-0.35	-0.26	-0.18
$\frac{\Delta C_A}{\Delta t}$ (ppm/min)		-0.071	-0.051	-0.035	-0.026	-0.018



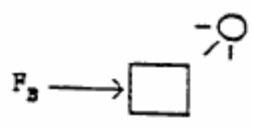


After plotting and differentiating by equal area

$-dC_A/dt$	0.082	0.061	0.042	0.030	0.0215	0.014
$\ln(-dC_A/dt)$	-2.501	-2.797	-3.170	-3.507	-3.840	-4.269
$\ln C_A$	0.896	0.554	0.207	-0.128	-0.478	-0.821

Using linear regression: $\alpha = 1.0$
 $\ln k = -3.3864 \rightarrow k = 0.0344 \text{ min}^{-1}$

(b)



$$\frac{dC_A}{dt} = Vr_A = F_B$$

$$r_A = -0.0344 \frac{\text{ppm}}{\text{min}} = -0.0344 \frac{\text{mg}}{\text{l min}} \text{ at } C_A = 1 \text{ ppm}$$

$$F_B = (25000 \text{ gal}) (0.0344 \frac{\text{mg}}{\text{l min}}) \frac{60 \text{ min}}{\text{hr}} \frac{1 \text{ g}}{1000 \text{ mg}} (\frac{3.7851}{453.6}) \frac{1}{\text{gal}} = 0.429 \frac{\text{lbm}}{\text{hr}}$$